

# **Safety Data Sheet**

# Better Chemistry. Better Business

ACID BRITE NCT Revised: 1/13/15

### **IDENTIFICATION**

Product Code: 2541050

Recommended use of the chemical and restrictions on use:Industrial applications

**Hubbard-Hall Inc.** 

563 South Leonard Street Waterbury, CT 06708 **Telephone**: 203-756-5521 **Fax number:** 203-756-9017

Emergency Phone Number **CHEMTREC:** 1 (800) 424-9300 **International:** 1 (703) 527-3887

## 2 HAZARDS IDENTIFICATION





Signal Word: DANGER

Hazard Category: Acute Toxicity-Oral Hazard Category 4

Eye Damage/Irritation Hazard Category 1 Skin Corrosion/Irritation Hazard Category 2 Acute Toxicity-Inhalation Hazard Category 4

Specific Target Organ Toxicity (Single Exposure) Hazard Category 3

Acute Aquatic Toxicity-Category 2

Hazard Statements: Harmful if swallowed or inhaled.

Causes skin irritation.

Causes serious eye damage. May cause respiratory irritation.

Toxic to aquatic life

**Prevention:** Wash skin thoroughly after handling.

Do not eat, drink or smoke when using this product.

Avoid breathing dust, fumes, gas, mist, vapors and sprays. Wear rubber gloves, goggles and chemical protective clothing.

Use only outdoors or in well ventilated area.

Keep only in original container.

Response: If swallowed: Rinse mouth. Do NOT induce vomiting.

If on skin (or hair): Take off immediately all contaminated clothing Rinse skin with water/shower .

Wash contaminated clothing before reuse.

If inhaled: Remove person to fresh air and keep comfortable for breathing. Call poison center/doctor if yo

Immediately call poison center or doctor and explain the type of exposure to the chemical(s) and provide the Specific treatment - refer to poison center or doctor for advice.

If in eyes: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to Absorb spillage to prevent material damage.

Storage: Store locked up.

Store in corrosive resistant high density polyethylene container.

Store in well ventilated place. Keep container tightly closed.

Disposal: Dispose of contents/container in accordance with local, regional, national, or international regulations.

### COMPOSITION INFORMATION

Chemical Name	Common Name And Synonyms	CAS No. and other Unique identifiers	Concentration %
Trade Secret	-	-	-

#### 4 FIRST AID

#### After Inhalation:

Remove from contaminated atmosphere. If breathing has ceased, clear the victim's airway and start mouth-to-mouth artificial respiration, which may be supplemented by the use of a bag-mask respirator, or manually triggered, oxygen supply capable of delivering 1 liter/second or more. If the victim is breathing, oxygen may be administered from a demand-type or continuous flow inhalor, preferably with a physician's advice. Contact a physician immediately.

### After Skin Contact:

Immediately remove contaminated clothing under a safety shower. Flush all affected areas with large amounts of water for 15 minutes. DO NOT attempt to neutralize with chemical agents. Obtain medical advice.

# After Eye Contact:

Immediately flush the eyes with large quantities of running water for 15 minutes. Hold the eyelids apart during the flushing to ensure rinsing of the entire surface of the eyelids with water. DO NOT attempt to neutralize with chemical agents. Obtain medical attention as soon as possible. Oils or ointments should not be used. Continue rinsing for an additional 15 minutes if the physician is not available.

# After Ingestion:

DO NOT induce vomiting. Immediately give large quantities of water or milk, if available. If vomiting does occur, give fluids again. Never give anything by mouth to an unconscious person. Call a physician or the nearest Poison Control Center.

#### Most Important Symptoms/Effects

### Delayed:

Will aggravate breathing disorders.

# FIRE FIGHTING MEASURES

Suitable and Unsuitable extinguishing media:

If involved in a fire, use water spray. Neuralize with soda ash or slaked lime.

Specific hazards arising from the chemical:

This product may release flammable hydrogen gas on contact with metal, which may significantly contribute to the risk of fire and explosion.

Special protective equipment and precautions for firefighter

In the event of a fire, wear full protective clothing and NIOSH approved self contained breathing apparatus with full facepiece operated in the pressure demand or other positive pressure mode. Structural firefighter's protective clothing is ineffective for fires involving this material Stay away from ends of tanks. Cool tanks with water spray until well after fire is out

# **6 ACCIDENTAL RELEASE MEASURES**

Personal Precautions, Protective Equipment, & Emergency Proc Prevent spilled product from drains, sewers, waterways and soil.

**Methods and Materials for** containment & cleaning up: Neutralize spill with soda ash or lime under good ventilation. For an interior (inside a closed space) spill be aware that the use of Soda Ash, Lime will eveolve heat and carbon

dioxide thus the need for ventilation.

If trained in accordance 29 CFR 1910.120, leaks should be stopped. Spills should be contained and cleaned immediately. Persons performing clean up work should wear adequate personal protective equipment and clothing. Spills and releases should be reported, if required, to the appropriate local, state and federal regulatory agencies.

# **HANDLING AND STORAGE**

Precautions for safe handling: Use in well ventilated area.

Avoid breathing dust, fumes, gas, mist, vapors and sprays.

Wear rubber protective gloves, chemical protective clothing, eye protective goggles and

face shield for face protection.

Keep only in original container.

Wash hands thoroughly after handling.

Do not get in eyes, or on skin, or on clothing.

Conditions for safe storage, inc any incompatibilities:

Keep container tightly closed.

Store locked up and away from incompatible chemicals.

Store in a well ventilated place. Keep cool.

# **EXPOSURE CONTROLS / PERSONAL PROTECTION**

Name	Std.	TWA-8hrs	STEL - 15 min.
Trade Secret	ACGIH	2 pm	

**ACGIH - American Control of Governmental Hygenists** OSHA - Occupational Safety and Health Administration

## **PHYSICAL AND CHEMICAL PROPERTIES**

Colorless to light yellow liquid Appearance:

Odor: Mild acidic odor

**Odor Threshold:** N/A PH: Below 1 **Melting Point/Freezing Point:** N/A **Initial Boiling Point and Boiling** N/A

Range:

**Evaporation Rate:** N/A

N/A Flammability (solid, gas): Upper/Lower flammability or

explosive limits:

N/A

Vapor Pressure: N/A

**Relative Density:** 1.14-1.15 @ 70 °F Complete in water Solubility (ies):

**Partition Coefficient;** 

N/A

n-octanol/water:

N/A **Auto-ignition Temperature:** N/A **Decomposition Temperature:** 

N/A Viscosity:

## 10 STABILITY AND REACTIVITY

**Chemical Stability:** Stable

**Conditions to Avoid:** Extremely reactive. Avoid contact with metal surfaces and oxidizing agents.

Chemically stable when properly contained and handled. It is a strong mineral acid and reacts **Incompatible Materials:** 

> with many metals and metal oxides and hydroxides to form equivalent metal chloride. It reacts with zeolites and other silicous compounds to form Hydrosilicic Acid; it reacts with carbonates to form Carbon Dioxide and water. It is oxidized by Oxygen or electrolysis to form Chlorine, a lethal poisonous gas. It reacts with alkaline compounds to form neutral salt. It is a hydrolyzing agent for carbohydrates, esters and other compounds. It's reaction with most metals will produce Hydrogen, an explosive gas. Violent reactions will result with acetic anhydride, 2-aminoethanol, ammonia hydroxide, calcium phosphide, chlorosulfonic acid, ethylene diamine, ethylene imine, oleum, perchloric acid, beta propiolactone, propylene oxide, sodium

hydroxide, uranium phosphide and vinyl acetate. This listing is not all inclusive.

**Hazardous Decomposition** 

Extreme heat may cause the product to decompose, producing toxic fumes which may include

chlorine compounds. Products:

### 11 TOXICOLOGICAL INFORMATION

Not established for this product **Oral Administration:** Inhalation: Not established for this product Not established for this product **Dermal administration:** Short term exposure: Irritation/ burns of skin and eyes.

Long term exposure to concentrated vapors may cause erosion of the teeth. Long term Long term exposure:

exposure seldom due to corrosive properties of the acid.

Numerical measures of

toxicity(such as toxicity

measurement) **Cancer Hazard:** 

Not listed by IARC, NTP, OSHA, ACGIH

Eyes, Skin, Inhalation, Ingestion **Routes of Exposure** 

Not known

# 12 ECOLOGICAL INFORMATION

Not Available Persistence and

Degradability:

No data available Abiotic degradability: Bioaccumulation potential: No data available

Soil/Sediment Result: Pronounced solubility and mobility

#### 13 DISPOSAL CONSIDERATION

Dispose of in accordance with local, state and federal regulations.

## 14 TRANSPORT INFORMATION

**UN Number:** 

NOT D.O.T REGULATED IN POLY CONTAINERS. **UN Proper Shipping Name:** 

Transport Hazard Class (es):

### 15 REGULATORY INFORMATION

HMIS: Health: 1 Flammability: 0 Reactivity: 0

# Sara Hazard Classification

SARA Title III Section 311 Categories: Immediate (Acute) Health Effects: Yes, Delayed (Chronic) Health Effects: No, Fire Hazard: No, Sudden Release of Pressure Hazard: No, Recativity Hazard:

No

# **16 OTHER INFORMATION**

No RoHS or REACH SVHC are contained in this product.

Disclaimer:

The information is based on our knowledge to date but does not constitute an assurance of product properties and does not imply a legal contractual relationship.